

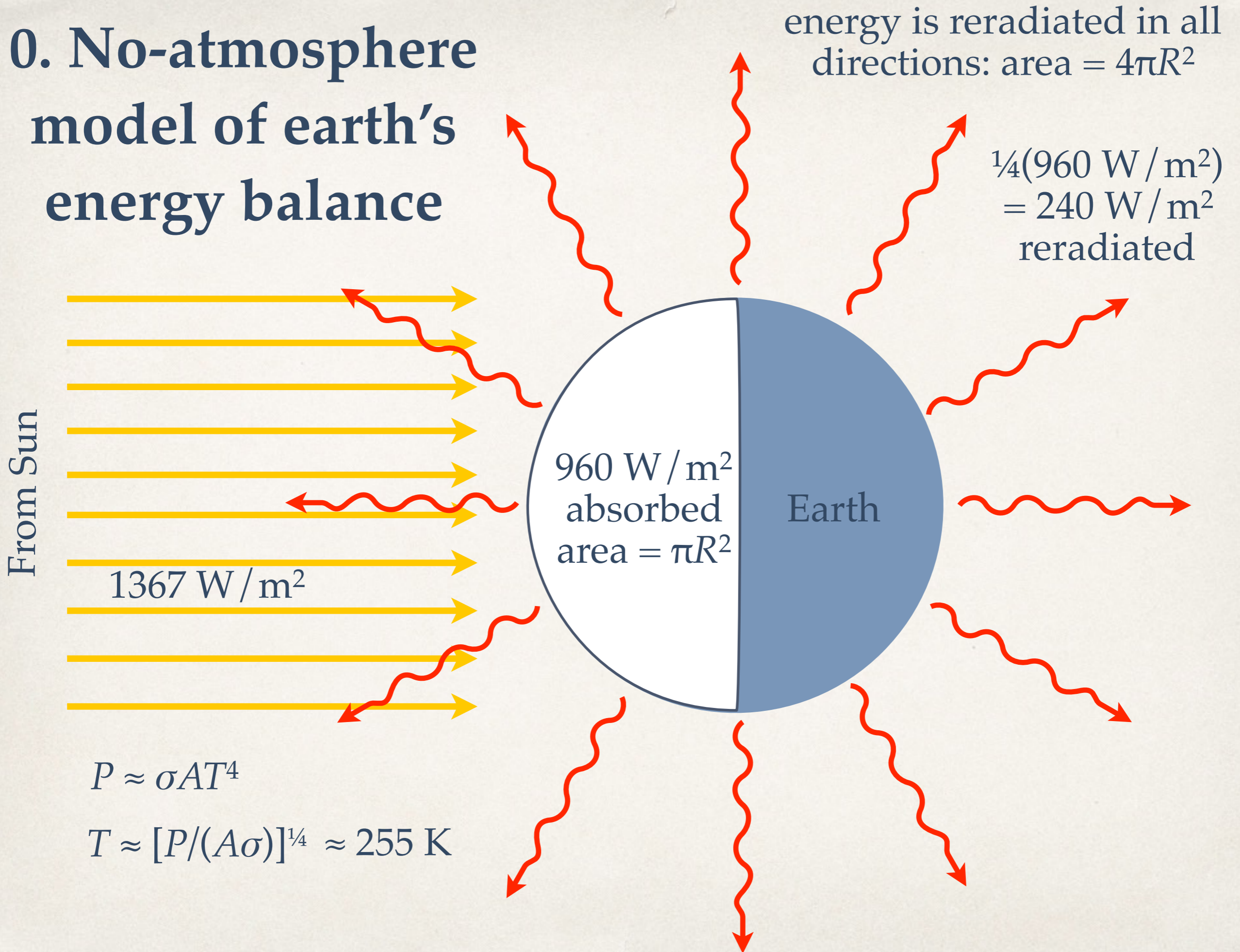
Climate Change in the Introductory Course

Chapter T10 in
Six Ideas 3/e!

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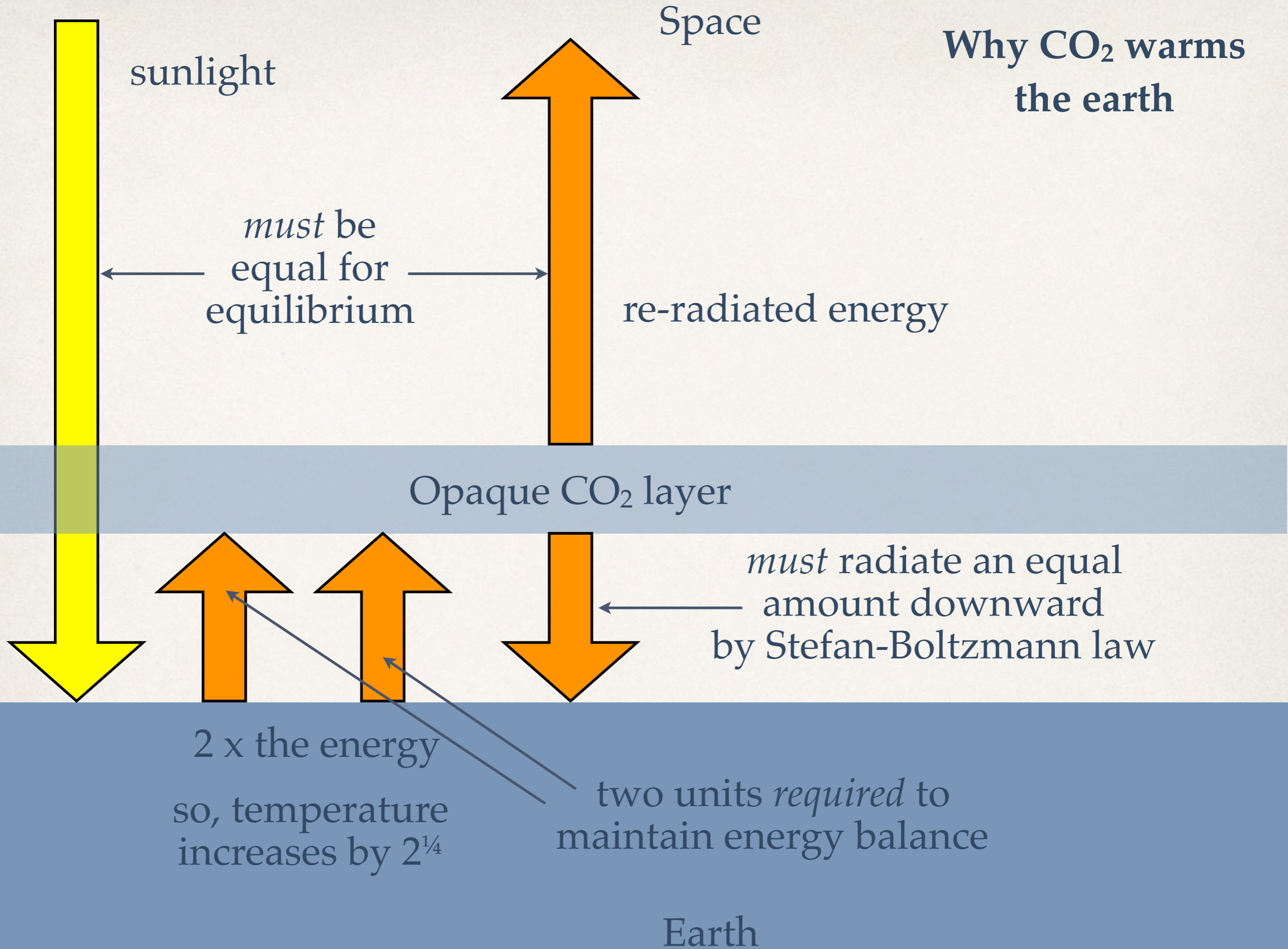
0. No-atmosphere model of earth's energy balance

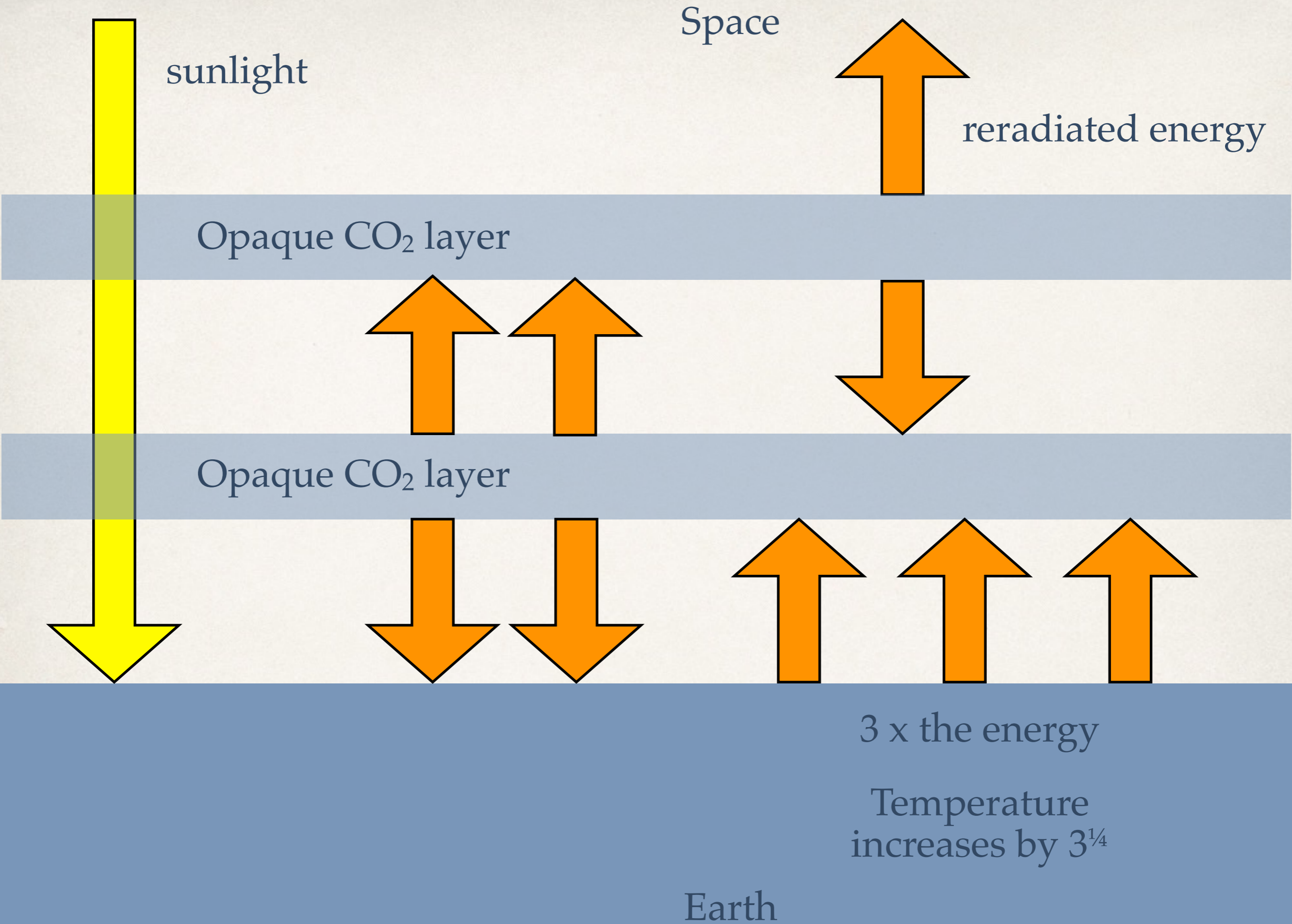


$$P \approx \sigma AT^4$$

$$T \approx [P/(A\sigma)]^{1/4} \approx 255 \text{ K}$$

Why CO₂ warms the earth





1. Pre-Industrial Model (for 280 ppm CO₂)

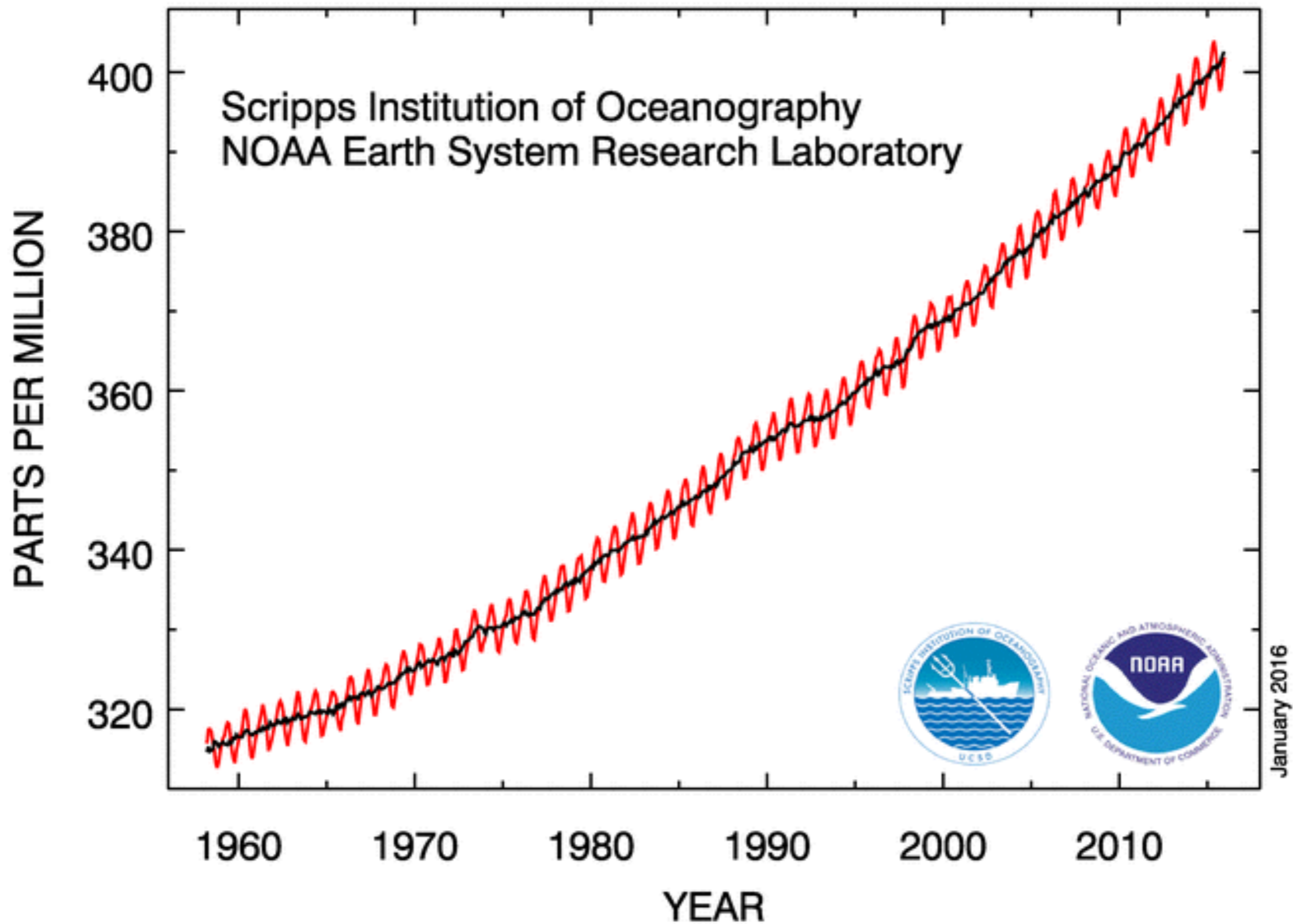
number of layers
↓

$$\text{So, } T = (255 \text{ K})(x + 1)^{1/4}$$

Let's *assume* this applies for non-integer x , but note that the true formula *should* match when x is an integer.

$$T = 14.0^\circ\text{C} = 287.0 \text{ K before 1900}$$
$$\Rightarrow x = 0.605 \quad \text{at 280 ppm CO}_2$$

Atmospheric CO₂ at Mauna Loa Observatory



2. Linear Increase Model

at 400 ppm CO₂, $x = (400 / 280)0.605 = 0.864$

$\Rightarrow T = (255 \text{ K})(1 + 0.864)^{1/4} = 298 \text{ K} = 25^\circ\text{C}$

$\Rightarrow T$ increases by 11°C (!)

**This is inconsistent with the the actual measured
(and not really disputed) rise of $\sim 0.8^\circ\text{C}$.¹**

3. Square Root Increase Model²

at 400 ppm CO₂, $x = (400 / 280)^{1/2} 0.605 = 0.723$

$$\Rightarrow T = (255 \text{ K})(1 + 0.723)^{1/4} = 292 \text{ K} = 19^\circ\text{C}$$

$\Rightarrow T$ increases by 5°C (!)

This is inconsistent with the the **actual measured**
(and not really disputed) rise of $\sim 0.8^\circ\text{C}$.¹

4. Adjust for CO₂ fraction

Pre-industrial CO₂ actually only contributes 27% of opacity (H₂O most of the rest).³ Now, $0.27(0.605) = 0.163$, so this is the effective thickness of the pre-industrial CO₂ layer and 0.442 is the thickness of the the other stuff \approx constant.

$$x = (400 / 280)^{1/2}(0.163) + 0.442 = 0.637$$

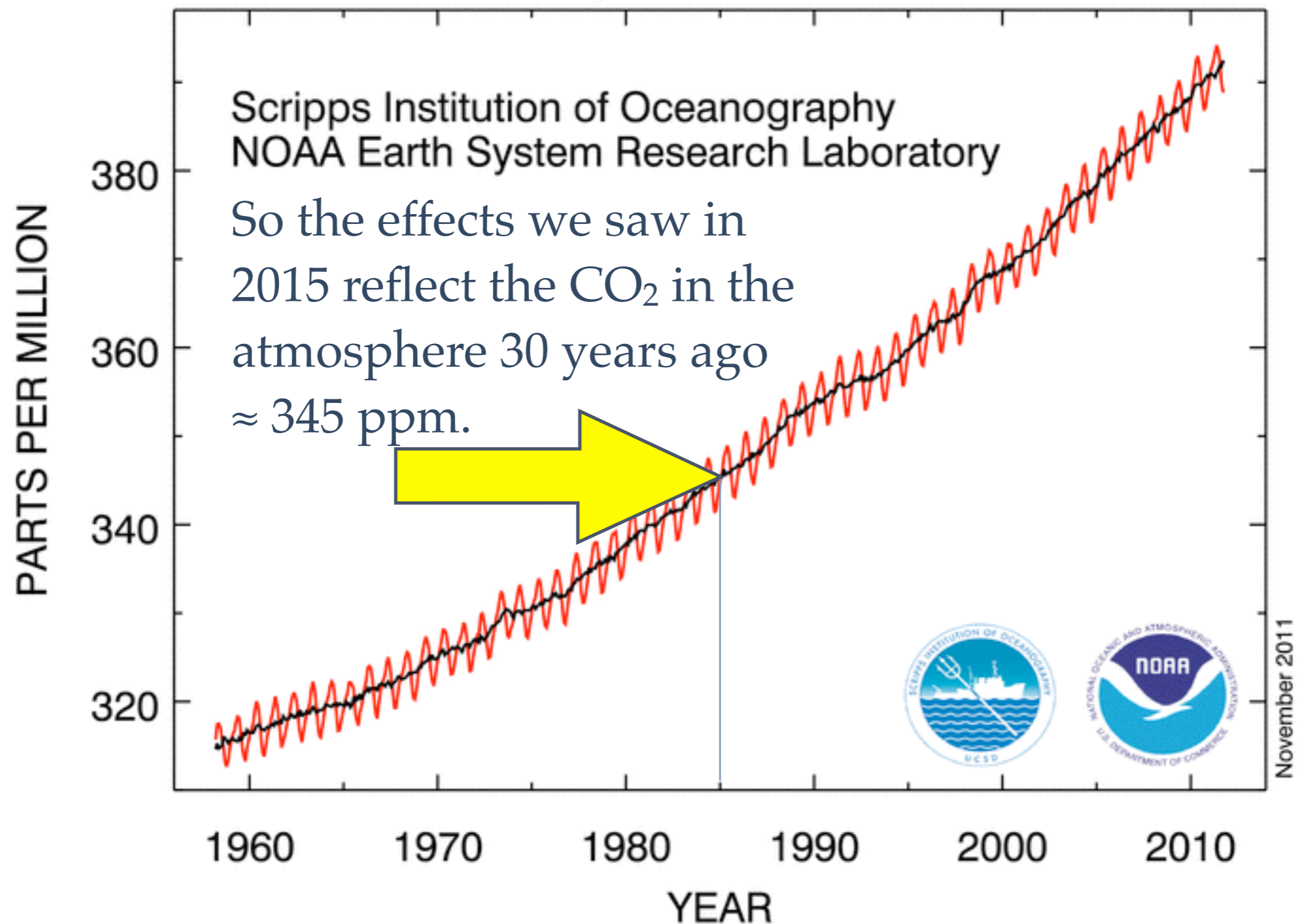
$$\Rightarrow T = (255 \text{ K})(x+1)^{1/4} = 288.4 \text{ K} = 15.4^\circ\text{C}$$

$$\Rightarrow \sim 1.4^\circ\text{C temperature increase}$$

5. Add Ocean Delay

- ❖ Model assumes equilibrium
- ❖ But Earth is currently *not* in equilibrium
- ❖ The ocean's large heat capacity delays equilibrium
- ❖ Simple model: effective delay is 30 years⁴

Atmospheric CO₂ at Mauna Loa Observatory



Calculating the temperature rise for 345 ppm:

$$x = (345 / 280)^{1/2}(0.163) + 0.442 = 0.623$$

$$\Rightarrow T = (255 \text{ K})(x+1)^{1/4} = 287.8 \text{ K} = 14.8^\circ\text{C}$$

$\Rightarrow \sim 0.8^\circ\text{C}$ temperature increase

This is basically spot on.

Endnotes

1. <http://www.ncdc.noaa.gov/sotc/global/2015/11/supplemental/page-2>

2. For more information about why the square root, go to <http://scienceofdoom.com/roadmap/co2/>

read CO₂ parts 3 and 4, and note that I am assuming that the “strong condition” applies (as claimed in the article) and that the “optical depth” is proportional to x (the number of layers).

3. Part 5 of the above discusses the 27% figure. (In general, scienceofdoom.com (in spite of the silly name) is a useful website for exploring the science of climate modeling at a significantly more sophisticated level than I am assuming here.)

4. I got this figure from a talk by climate scientist James Hansen. See also <http://arxiv.org/abs/1307.6821> and <http://meteora.ucsd.edu/cap/pdf/Hansen-04-29-05.pdf>

Chapter T10 in
Six Ideas That Shaped Physics, 3rd edition
(McGraw-Hill, available February 2016)

<http://www.physics.pomona.edu/sixideas/>

(I have posted this talk there)